EFFECT OF MAGNET SCRAP SIZE ON THE EXTRACTION BEHAVIOR OF HEAVY RARE EARTH ELEMENTS
BY LIQUID METAL EXTRACTION

Liquid metal extraction (LME) process results in 100% neodymium (Nd) extraction but the highest extraction efficiency reported for Dysprosium (Dy) so far is 74%. Oxidation of Dy is the major limiting factor for incomplete Dy extraction. In order to enhance the extraction efficiency and to further investigate the limiting factors for incomplete extraction, experiments were carried out on six different particle sizes of under 200 μm, 200-300 μm, 300-700 μm, 700-1000 μm and over 2000 μm at 900℃ with magnesium-to-magnet scrap ratio of 15:1 for 6, 24 and 48 hours, respectively. This research identified Dy2Fe17 in addition to Dy2O3 phase to be responsible for incomplete extraction. The relationship between Dy2Fe17 and Dy2O3 phase was investigated, and the overall extraction efficiency of Dy was enhanced to 97%.

Keywords: Liquid metal extraction, Rare earth elements, Recycling, Oxidation, Magnet scrap

1. Introduction

Neodymium (Nd) based permanent magnets store the highest magnetic energy per unit volume as compared to other magnets [1]. Dysprosium (Dy), a heavy rare earth element (HREE) and a critical material [2], is added to enhance the temperature stability of the magnet for improving its ability to perform at elevated temperatures in Nd based permanent magnets [3]. Besides being key component of renewable energy technologies such as wind turbines, Rare earth (RE) magnets are used in Eco-friendly vehicles, electronic devices, alloy elements etc. [4]. With the world’s shift towards greener economy, the demand of REEs have tremendously increased but the skewed dynamics of the RE market poses supply-demand risk which need to be urgently addressed.

Substantial researches focused on the recycling of Nd and Dy from permanent magnets using environmentally friendly techniques such as direct melting, electro-slag refining, glass slag method and liquid metal extraction (LME) have been carried out in the past. Process parameters such as calculations of diffusion coefficients, sequences of extraction mechanisms and factors for optimization of extraction process were reported in these studies [5-8]. However, effect of scrap size on the extraction ratio and diffusion behavior have not been studied in detail. A similar study on effect of scrap size from Nd-Fe-B magnet, which did not consider Dy, reported the maximum extraction for Nd to be 96.5% [9]. From other studies, the highest extraction efficiency reported for Dy in LME process so far is of 74% [8].

REs available for recycling are present in different shapes and sizes, therefore, for practical implications six different scrap sizes of different shapes were selected for experimental purposes to gauge the effect of scrap size on the extraction efficiency by the LME process with Mg as an extraction agent. Moreover, the effect of oxidation and the phase transformations during the process were also investigated.

2. Experimental

RE magnets containing ~30wt.% RE were obtained from JAHWA electronic Co. Ltd. for experimental purposes. The stoichiometric composition of these rare earth magnets as analyzed by XRF is shown in Table 1.

These magnets were pulverized using jaw crusher into six different particle sizes of under 200 μm, 200-300 μm, 300-700 μm, 700-1000 μm, 1000-2000 μm and over 2000 μm.
Oxygen content of each sample was measured by Nitrogen-Oxygen-Hydrogen (NOH) analyzer.

Magnesium of 99% purity was procured from JC company. Magnesium was selected as an extracting agent because it selectively reacts with Nd and Dy in the magnet and it hardly forms any intermetallic compound with Fe [5]. Magnesium was grind to remove the oxidation layer and to fit inside the 1 cm × 1 cm × 10 cm Fe crucible. The LME reactions were carried out in an induction furnace for reaction times of 6h, 24 h and 48 hours at a fixed temperature of 900℃ with Mg: Magnet ratio of 15:1. Vacuum was achieved inside the induction furnace up to a level of 1 × 10^{-2} torr before high purity Ar gas was introduced in the chamber for creating an inert environment in order to prevent oxidation. The heating rate was kept under 10℃/min and the sample was let to cool naturally after the completion of reaction time.

The crucible was cut by diamond blade for observing the cross-section of the (Nd,Dy)FeB-Mg coupled sample. The extraction efficiency was calculated using XRF while the morphology of the cross-section was observed by FE-SEM, and the phases were identified by XRD.

### 3. Results and discussion

The results of oxygen content of each scrap size are shown in Table 2.

Higher oxidation concentration was reported in smaller scrap sizes which can be associated to the increase in the surface to volume ratio in smaller particle sizes. RE oxidation takes place easily, especially during the grinding process in which the magnet is exposed to atmosphere at a high thermal energy state. As per the microstructural analysis of the cross-sectioned RE magnet scrap-Mg sample after the reaction, the sample can be divided into three zones i.e. unaffected magnet zone, reaction zone and Mg zone. The region where REs had diffused out of the magnet into Mg to form intermetallic compounds was labelled as reaction zone, hence it is the zone of most importance from scientific point of view.

In order to observe how the scrap size and oxidation content influenced the phase morphology and extraction behavior, the BSE images of the reaction zone of the six different scrap sizes at 48 hours are shown in Fig. 1.

Mg infiltrated the magnet forming black ligaments, and it can be observed from the size of the ligaments in the images that Mg infiltrated more in smaller scrap sizes because of the larger reaction surface. A grey zone (Fe) was formed as a result of the reaction between RE2Fe14B and Mg. Moreover, nano size particles were observed in all scrap sizes. As shown in Fig. 1(a) and (f) these particles were divided into two categories, (#1) and (#2), depending on their location. Nano size particles in the ligaments were termed as (#1) and nano size particles in Fe-rich area as (#2). The amount of nano size particle (#1) decrease as the size of the scrap increases, whereas, nano-size white particles (#2) increase with the increase in scrap size.

### TABLE 1

<table>
<thead>
<tr>
<th>Elements</th>
<th>Nd (wt.%)</th>
<th>Dy (wt.%)</th>
<th>Fe (wt.%)</th>
<th>Minor (wt.%)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Rare earth magnet</td>
<td>25.8</td>
<td>3.5</td>
<td>68.98</td>
<td>1.72</td>
</tr>
</tbody>
</table>

### TABLE 2

<table>
<thead>
<tr>
<th>Powder size</th>
<th>Under 200 μm</th>
<th>200-300 μm</th>
<th>300-710 μm</th>
<th>710-1000 μm</th>
<th>1000-2000 μm</th>
<th>Over 2000 μm</th>
</tr>
</thead>
<tbody>
<tr>
<td>Oxygen (wt.%)</td>
<td>0.2612</td>
<td>0.2540</td>
<td>0.1773</td>
<td>0.1720</td>
<td>0.1621</td>
<td>0.1134</td>
</tr>
</tbody>
</table>

Fig. 1. BSE images of the reaction zone in magnet after 48 hours at 900℃ (a) under 200 μm, (b) 200-300 μm, (c) 300-700 μm, (d) 700-1000 μm, (e) 1000-2000 μm, and (f) over 2000 μm
The phases identified in the reaction zone are highlighted in Fig. 2(a) from the XRD results shown in Fig. 2(b). As reaction proceeds, the RE₂Fe₁₄B, RE-rich phase and RE-oxide in magnet transform into Fe, REMg₁₂, RE₅Mg₄₁, Dy₂Fe₁₇ and Dy₂O₃ phase in the reaction zone. Nano size particles (#1) were identified as Dy₂O₃ and nano-size particles (#2) as Dy₂Fe₁₇ by SEM-EDS and XRD.

The infiltration of Mg and the increase in oxygen content can be directly linked with the extraction efficiency. For observing this linkage, the extraction efficiencies of the six different particle sizes after a reaction time of 6, 12 and 48 hours are shown in Fig. 3.

As can be seen from Fig. 3 that 100% Nd extraction was achieved irrespective of the particle size and on all reaction times above 6 hours. However, in case of Dy, the reaction time had a significant effect on the extraction efficiency as it increased in all the particle sizes with an increase in reaction time. The overall maximum efficiency of 64, 81 and 97% against 6, 24 and 48 hours, respectively, was achieved in the particle size of over 2000 μm.

Incomplete decomposition of Dy₂Fe₁₇ and incomplete reduction of Dy₂O₃ were identified as the reasons for Dy extraction efficiencies of less than 100%. Incomplete reduction of Dy oxide has also been reported by a previous research [8], hence, solution to remove Dy₂Fe₁₇ and Dy₂O₃ is needed for complete Dy extraction. The reason for formation of Dy₂Fe₁₇ can be explained by the negative heat of mixing value of Dy and Fe, which means that Dy has a stronger affinity for Fe as compared to Nd during reaction [10], whereas, Dy₂O₃ forms due to the presence of oxygen content in magnet.

Moreover, a correlation between the oxidation content of each particle size and respective extraction efficiency was observed. The higher the oxidation of Dy into Dy₂O₃, the lesser was the extraction. For confirming the effect of Dy phase transformation on extraction efficiency, the Dy related phase fractions in each particle size at 6, 24 and 48 hours were measured, and the results are displayed in Table 3.

### TABLE 3

<table>
<thead>
<tr>
<th>Powder size</th>
<th>Dy₂O₃ phase fraction (%)</th>
<th>Dy₂Fe₁₇ phase fraction (%)</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td>6 hours</td>
<td>24 hours</td>
</tr>
<tr>
<td>Under 200 μm</td>
<td>1.58</td>
<td>0.95</td>
</tr>
<tr>
<td>200-300 μm</td>
<td>1.49</td>
<td>0.88</td>
</tr>
<tr>
<td>300-710 μm</td>
<td>1.30</td>
<td>0.68</td>
</tr>
<tr>
<td>710-1000 μm</td>
<td>1.26</td>
<td>0.66</td>
</tr>
<tr>
<td>1000-2000 μm</td>
<td>1.15</td>
<td>0.60</td>
</tr>
<tr>
<td>Over 2000 μm</td>
<td>0.91</td>
<td>0.48</td>
</tr>
</tbody>
</table>
The results show a trend of decreasing phase fraction of Dy$_2$O$_3$ with increasing particle size whereas it is the opposite for Dy$_2$Fe$_{17}$ in all the three experimental times. This implies that in particles with lesser oxygen content Dysprosium preferentially reacted with Iron as compared to Oxygen.

This change in Dy-phase fractions across the particle sizes effects the extraction efficiency of Dy i.e. greater the decrease in Dy$_2$O$_3$ phase, higher is the Dy extraction efficiency. It can be seen that the extraction rate is highest in the biggest particle size for all three timings. More importantly, the percentage of Dy$_2$Fe$_{17}$ phase is larger compared to Dy$_2$O$_3$ phase in all the particle sizes but percentage of Dy is higher in Dy$_2$O$_3$ than Dy$_2$Fe$_{17}$, therefore, it is logical to conclude that Dy$_2$O$_3$ has a greater effect on extraction efficiency than Dy$_2$Fe$_{17}$ phase.

4. Conclusions

Effect of scrap size on the recycling efficiency of RE permanent magnets in the LME process was observed. Oxygen content, which increased with decreasing particle size, was found to have a significant effect on the extraction rate. Consequently, the extraction efficiency increased with increasing particle size and increasing time. The highest extraction rate of 97% Dy was achieved in the particle size of over 2000 μm for a reaction time of 48 hours. Un-extracted Dy remained in the form of Dy$_2$O$_3$ and Dy$_2$Fe$_{17}$ phase. The phase fraction percentage of Dy$_2$O$_3$ decreased while Dy$_2$Fe$_{17}$ phase increased with increasing particle size, making Dy$_2$Fe$_{17}$ the more dominant among the two with higher phase fraction in larger particle size powders. However, the percentage of Dy in Dy$_2$Fe$_{17}$ phase is low as compared to Dy$_2$O$_3$ because of which Dy$_2$O$_3$ has a greater effect on Dy extraction efficiency than Dy$_2$Fe$_{17}$. In order to achieve 100% extraction efficiency for Dy, complete reduction of Dy$_2$O$_3$ and decomposition of Dy$_2$Fe$_{17}$ is required, where reduction of Dy$_2$O$_3$ remains to be the major problem. Complete reduction of Dy$_2$O$_3$ is an important topic for future research, and one possibility of achieving that is by using catalyst.

Acknowledgments

This research was supported by a grant from project of development of environment friendly pyrometallurgy process for high purity HREE and materialization (Project number: 20000970) by Korea evaluation Institute of Industrial Technology (KEIT) in Republic of Korea

REFERENCES