Thermodynamic analysis of cycle arrangements of the coal-fired thermal power plants with carbon capture

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Abstract The electricity production by combustion of organic fuels, especially coal, increases the atmospheric CO$_2$ content, which contributes to global warming. The greenhouse gas emissions by the power production industry may be reduced by the application of CO$_2$ capture and storage systems, but it remarkably decreases the thermal power plant (TPP) efficiency because of the considerable increase of the auxiliary electricity requirements. This paper describes the thermodynamic analysis of a combined cycle TPP with coal gasification and preliminary carbon dioxide capture from the syngas. Utilization of the heat produced in the fuel preparation increases the TPP net efficiency from 42.3% to 47.2%. Moreover, the analysis included the combined cycle power plant with coal gasification and the CO$_2$ capture from the heat recovery steam generator exhaust gas, and the oxy-fuel combustion power cycle with coal gasification. The coal-fired combined cycle power plant efficiency with the preliminary CO$_2$ capture from syngas is 0.6% higher than that of the CO$_2$ capture after combustion and 9.9% higher than that with the oxy-fuel combustion and further CO$_2$ capture. The specific CO$_2$ emissions are equal to 103 g/kWh for the case of CO$_2$ capture from syngas, 90 g/kWh for the case of CO$_2$ capture from the exhaust gas and 9 g/kWh for the case of oxy-fuel combustion.

Keywords: Combined cycle power plant; Carbon capture and storage system; Pre-combustion capture; Post-combustion capture; Oxy-fuel combustion

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Abbreviations

ASU – air separation unit
CCS – capture and storage
HRSG – heat recovery steam generator
IGCC – integrated gasification combined cycle
MEA – monoethanol amine
TPP – thermal power plant

1 Introduction

The carbon dioxide (CO$_2$) atmospheric content is about 410 ppm [1] and has grown by 10% in the last two decades [2]. The carbon dioxide leading producers are China, the USA, India, and Russia [3].

Nowadays reduction of greenhouse gases emission is one of the most important tasks. The 2015 Paris climate agreement requires carbon dioxide emissions to reach zero level by 2050 [4,5]. The world power industry mostly produces heat and electricity but its carbon dioxide emissions are about a quarter of the world amount [6]. It is proposed to reduce the power industry emissions by the introduction of renewable power sources, nuclear power plants, and low-carbon thermal power plant (TPP) technology.

Introduction of the renewable power sources is limited by the low density of energy flow and its irregularity. Nuclear power plant production is related to the risk of accidents that may cause radiation pollution. These factors and the wide use of TPP make prospects for carbon capture and storage (CCS) technology. On the other side, this technology has high power consumption and requires large operation and investment outlays. Thus, the wide CCS introduction requires searches for the most efficient and environmentally safe version of the TPP cycle arrangement that provides the required emission reduction combined with the acceptable fuel consumption and moderate cost.

The CCS concept may be presented in three directions (Fig. 1), the pre-combustion carbon dioxide capture, the post-combustion capture and the oxy-fuel combustion capture. Versions of these technologies provide an 85–99% capture degree [7,8].

Each of the technology versions may be split into a few sub-technologies. The pre-combustion capture involves the absorption by physical or chemical reactants. In the physical absorption technology, the captured CO$_2$ price is 25–30 U.S. dollar per ton which is remarkably cheaper than the 40–50 U.S. dollar per ton in the chemical absorption [9], because after the CO$_2$ separa-
tion the absorbent may be completely separated from the absorbate. In this technology, the syngas is preliminarily enriched where its hydrogen content in the fuel mixture is increased and the carbon monoxide is transformed into carbon dioxide. Then the CO₂ is separated by a physical absorbent and is supplied to the CCS [10, 11]. This technology may be used only in the pre-combustion capture when the gas fuel produced by the natural gas conversion or by gasification has a pressure above 8 bar [12].

The post-combustion technology may employ the membrane [13–15] or cryogenic [16,17] separation, or the oxy-fuel combustion capture [18–20], or the chemical looping combustion [21–24], or the chemical reagent absorption like in the pre-combustion capture technology. These technologies may be used for modification of the existing steam turbine and combined cycle power plants, or the new TPP construction. Table 1 summarizes the main performance of the post-combustion carbon dioxide capture. It shows that the widely used CO₂ chemical absorption from exhaust gas is one of the most efficient methods.

Table 1: Comparison of the post-combustion carbon capture technology characteristics.

<table>
<thead>
<tr>
<th>Parameter</th>
<th>Chemical absorption</th>
<th>Membrane separation</th>
<th>Cryogenic separation</th>
<th>Chemical looping combustion</th>
</tr>
</thead>
<tbody>
<tr>
<td>Specific energy consumption for CO₂ capture, MJ/kg</td>
<td>0.59–2.83</td>
<td>0.6–6.0</td>
<td>0.49–3.4</td>
<td>3.8–4.9</td>
</tr>
<tr>
<td>CO₂ capture rate, %</td>
<td>85–90</td>
<td>80–90</td>
<td>&gt;95</td>
<td>&gt;95</td>
</tr>
</tbody>
</table>

The advantages of the carbon dioxide capture with monoethanol amine (MEA) are the high capture degree of 85–90% in a wide range of its partial pressures, the high chemical stability and the high reaction capability. The technology shortages are the absorbent losses that need a refill, the
surface’s corrosion and the high price. Despite these shortages, the MEA capture technology is the best approved and relatively cheap, so it is widely used [25–29].

Another advanced direction to the CO₂ emission reduction is the transition to oxy-fuel combustion power cycles. The organic fuel combustion in the oxygen area produces a two-component working fluid which allows its split into carbon dioxide and water vapor through the vapor condensing [30]. In this method, the CO₂ capture degree is about 99%.

Numerous papers compare the influence of different capture systems upon the power plant efficiency. Specifically paper [12] discloses the thermal efficiency comparison results of a natural gas-fired combined cycle power plant equipped with the post-combustion CCS system with a coal-fired steam turbine power plant equipped with the CCS system and the coal-fired integrated gasification combined cycle (IGCC) equipped with the pre-combustion CCS system. The conclusion is that the most efficient coal-fired power production is reached in IGCC with the physical absorption of carbon dioxide. However, among the reviewed TPP versions there were neither coal-fired IGCC with the post-combustion carbon capture nor oxy-fuel combustion power cycle with the carbon dioxide working fluid. In turn, the paper [31] reviews all three methods in a combined cycle TPP with coal gasification and it mentions that oxy-fuel combustion requires new solutions in the power production equipment design. It is worth mentioning that the published papers don’t compare the most efficient coal-fired TPP flow concepts with different CCS systems at currently available working fluid parameters. Therefore, in this paper the comparison is carried out for the most efficient versions of the coal-fired TPP with low greenhouse gases emission as the following:

- IGCC with pre-combustion capture with the CO₂ physical absorption separation from the coal gasification syngas;
- IGCC with post-combustion capture with the CO₂ chemical absorption from flue gas with the monomethanol amine solvent;
- oxy-fuel combustion power plant based on the Allam cycle that is one of the most efficient oxy-fuel combustion power cycles [20,32,33].

The additional thermodynamic analysis discloses the utilization of the heat produced in the syngas preparation before its firing in IGCC.
2 Research object

The combined cycle power plants with steam-oxygen coal gasification equipped with CCS systems are studied in this work. The power plants in this study differ by the CO₂ capture technology.

The first unit (Fig. 2a) is a coal-fired IGCC. Gasification block 3 transforms coal into syngas. It consists of the gasifier and the ash interception. The processing oxygen is produced by the air separation unit (ASU) 1 from the atmosphere air and compressed up to gasifier pressure in the inter-cooled oxygen compressor 2. The processing steam is taken from the steam turbine 14. Then the syngas is fired in the gas turbine combustion chamber 5. The air compressor 4 compresses air. Some air is supplied to the gas turbine 6 cooling. The gas turbine 6 is mounted together with the power generator 7 on a single shaft. Then the gas turbine exhaust enters the double-pressure heat recovery steam generator (HRSG) where it transfers its heat to the steam turbine cycle through heat exchanger surfaces. Then the block 21 captures the exhaust gas CO₂ by the chemical absorption method. The CO₂ is compressed in compressor 22 and sent to its storage.

In the steam turbine cycle, downstream the condenser 16 the condensate pump 19 supplies water to the condensate gas heater 13. Upstream the heater a part of the heated water is recirculated with the recirculation pump 20. Then the water passes the deaerator 15 and is split into two the low- and high-pressure flows by the feedwater pumps 17 and 18. The low-pressure feedwater sequentially passes the low-pressure vaporizer 12, the steam superheater 11 and enters the steam turbine mixing chamber 14 except the deaeration flow. In the mixer chamber, the steam is mixed with the high-pressure steam.

The high-pressure feed water is supplied to the economizer 10, high-pressure vaporizer 9 and steam superheater 8 like in the low-pressure circuit. The high-pressure steam enters the steam turbine 14 where it produces power. The turbine exhaust steam enters the condenser 16.

The second version (Fig. 2b) shows an IGCC with pre-combustion capture. This cycle arrangement differs from that in Fig. 2a. After the gasification block, the syngas is preliminarily enriched in the water gas shift reactor 23. Then in the physical absorption column 24, it is cleaned from the carbon dioxide. The carbon dioxide is cooled in cooler 26, compressed in compressor 22 and sent to the storage. From the enriched and cleaned syngas water is separated by condensing in the cooler 26 then it is supplied to the combustion chamber 5.
The third unit (Fig. 2c) is a coal-fired oxy-fuel combustion power plant. Gasification block 3 is supplied with coal, oxygen and steam. The syngas is cooled in cooler 26 for a more efficient compression, then compressed in the fuel compressor 25 and supplied to the combustion chamber 5. The syngas is fired in the almost pure oxygen produced by the ASU 1. The combustion heat is transferred to the CO₂ flow passing the multi-flow regenerator 28. Downstream of the combustion chamber 5 the working fluid is supplied to the carbon dioxide turbine 29 where it produces power. The turbine 29 exhaust flow enters the multi-flow regenerator 28 where it transfers its heat to the CO₂ flow, the turbine 29 coolant flow and the CO₂ with O₂ mixture flow. Also, the regenerator utilizes the low-potential heat of the ASU 1 outlet air flow. After regenerator 28, its exhaust is supplied
to cooler-separator 30 where the working fluid water vapor is condensed. Then the CO$_2$ flow is split into two flows, the main part is compressed in the multi-stage compressor 31 and the residual part is compressed in compressor 22 and stored.

After the multi-stage compressor 31, the CO$_2$ flow is split into two equal parts. One of them is mixed with oxygen. After the carbon dioxide compressor 32, the flow is again split into two parts. The smaller part works as the turbine 29 coolant and is heated in the regenerator 28. In the regenerator the main CO$_2$ flow with the CO$_2$ and O$_2$ mixture are also heated. The oxygen-carbon dioxide compressor 33 compresses the mixture up to the maximum temperature acceptable for the regenerator temperature drop. The hot flows that enter combustion chamber 5 from regenerator 28 reduce the fuel flow needed for the working fluid heating up to the high temperature.

Table 2 presents the main parameters for modeling coal-fired power plants with coal gasification and CCS system. The fuel composition is taken according to [34]. The gas turbine inlet temperature of 1700°C corresponds to the advanced developments [35]. The oxy-fuel combustion power cycle parameters are taken according to [20].

<table>
<thead>
<tr>
<th>Parameter</th>
<th>Unit</th>
<th>Value</th>
</tr>
</thead>
<tbody>
<tr>
<td>Fuel type</td>
<td>–</td>
<td>coal</td>
</tr>
<tr>
<td>Fuel composition:</td>
<td></td>
<td></td>
</tr>
<tr>
<td>moisture (a.r.)</td>
<td>%</td>
<td>8.10</td>
</tr>
<tr>
<td>ash</td>
<td>%</td>
<td>14.19</td>
</tr>
<tr>
<td>carbon</td>
<td>%</td>
<td>72.04</td>
</tr>
<tr>
<td>hydrogen</td>
<td>%</td>
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</tr>
<tr>
<td>nitrogen</td>
<td>%</td>
<td>1.67</td>
</tr>
<tr>
<td>oxygen</td>
<td>%</td>
<td>7.36</td>
</tr>
<tr>
<td>sulphur</td>
<td>%</td>
<td>0.65</td>
</tr>
<tr>
<td>chlorine</td>
<td>%</td>
<td>0.01</td>
</tr>
<tr>
<td>Volatile matter (dry)</td>
<td>%</td>
<td>28.51</td>
</tr>
<tr>
<td>Oxygen purity</td>
<td>%</td>
<td>95.6</td>
</tr>
<tr>
<td>Gasifier pressure</td>
<td>MPa</td>
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</tr>
<tr>
<td>Carbon storage pressure</td>
<td>MPa</td>
<td>10</td>
</tr>
<tr>
<td>Internal turbine/compressor/pump efficiency</td>
<td>%</td>
<td>89/88/85</td>
</tr>
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</table>
Table 2 [cont.]

<table>
<thead>
<tr>
<th>Parameter</th>
<th>Unit</th>
<th>Value</th>
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<tbody>
<tr>
<td>Mechanical, electric motor, power generator, heat transport efficiency</td>
<td>%</td>
<td>99</td>
</tr>
<tr>
<td>IGCC’s/oxy-fuel combustion power plant’s turbine inlet temperature</td>
<td>◦C</td>
<td>1700/1100</td>
</tr>
<tr>
<td>Coolant flow rate of IGCC / oxy-fuel combustion power plant turbine</td>
<td>%</td>
<td>13.6/9.4</td>
</tr>
<tr>
<td>IGCC’s/oxy-fuel combustion power plant’s turbine inlet pressure</td>
<td>MPa</td>
<td>3/30</td>
</tr>
<tr>
<td>IGCC’s/oxy-fuel combustion power plant’s turbine outlet pressure</td>
<td>MPa</td>
<td>0.1013/3</td>
</tr>
<tr>
<td>High/low pressure of IGCC steam</td>
<td>MPa</td>
<td>8.55/0.7</td>
</tr>
<tr>
<td>Minimum temperature drop of IGCC high-/low-pressure heaters</td>
<td>◦C</td>
<td>20/20</td>
</tr>
<tr>
<td>IGCC condenser pressure</td>
<td>kPa</td>
<td>4</td>
</tr>
<tr>
<td>Condensate temperature at the inlet to the condensate gas heater of the</td>
<td>◦C</td>
<td>60</td>
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<tr>
<td>IGCC HRSG</td>
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<td></td>
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<tr>
<td>IGCC deaerator pressure</td>
<td>MPa</td>
<td>0.45</td>
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<tr>
<td>Minimum temperature drop of the multi-flow regenerator of the oxy-fuel</td>
<td>◦C</td>
<td>5</td>
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<tr>
<td>combustion power plant</td>
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</table>

3 Modeling approach

The thermodynamic analysis of TPP with coal gasification is carried out with the AspenONE Plus calculation models that consist of elements with the mass and energy transition between them. The elements models are the following:

- air separation unit that produces high purity oxygen, its modeling methodology is described in [36],
- gasification block with a steam-oxygen blast that transforms coal into syngas [37],
- combined cycle power unit producing electricity,
- oxy-fuel combustion power unit producing electricity,
- carbon capture and storage system.

The models of the IGCC with CCS system involves the following assumptions:

- at the gas generator outlet the ash and syngas have equal temperatures;
- the syngas combustion in the combustion chamber is stoichiometric.
Figure 3 presents the mass and energy flow exchange between the main blocks of the mathematical model elements as the following:

- IGCC with post-combustion carbon capture (Fig. 3a);
- IGCC with pre-combustion carbon capture (case 1), the heat produced in the fuel preliminary preparation system is not utilized (Fig. 3b);
- IGCC with pre-combustion carbon capture (case 2), the heat produced in the water gas shift reactor is utilized in the HRSG (Fig. 3c);
- IGCC with pre-combustion carbon capture (case 3), the heat produced in the water gas shift reactor and the syngas cooler is utilized in the HRSG (Fig. 3d);
- IGCC with pre-combustion carbon capture (case 4), the heat produced in the water gas shift reactor and the syngas and carbon dioxide coolers is utilized in the HRSG (Fig. 3e);
- coal-fired oxy-fuel combustion power plant (Fig. 3f).
The mathematical model of the dioxide carbon capture in the IGCC with post-combustion carbon capture by chemical absorption (Fig. 3a) involves two reactions. The first reaction between MEA and carbon dioxide is exo-thermal and passes at 40–60°C

$$\text{CO}_2 + 2\text{RNH}_2 + 2\text{H}_2\text{O} \leftrightarrow (\text{RNH}_3)\text{C}_3 + 66.15 \frac{\text{kJ}}{\text{kg}}. \quad (1)$$

The second reaction is the endothermal desorption, absorbent regeneration reaction. The absorbed carbon dioxide is emitted at 110–120°C

$$\text{CO}_2 + (\text{RNH}_3)\text{C}_3 + \text{H}_2\text{O} \leftrightarrow 2\text{RNH}_3\text{HCO}_3 - 66.15 \frac{\text{kJ}}{\text{kg}}. \quad (2)$$

Here the carbon dioxide capture degree is 90%.
In the analysis of the IGCC with pre-combustion carbon capture by physical absorption (Fig. 3b–3e), a process in water gas shift reactor described by the equation of the water gas shift exothermal reaction

\[
\text{CO} + \text{H}_2\text{O} \leftrightarrow \text{CO}_2 + \text{H}_2 + 41.2 \ \frac{\text{kJ}}{\text{mol}}.
\]

The next stage is the physical carbon dioxide absorption by the Selexol solvent from the hydrogen-enriched syngas. The Selexol is a mixture of various dimethyl ethers of polyethylene glycol \(\text{CH}_3\text{O(C}_2\text{H}_4\text{O)}_n\text{CH}_3\), where \(n\) is in the range from 2 to 9. The degree of carbon dioxide capture with the Selexol solvent is 95% [38]. Then the syngas is cooled to condense the ballast water vapor and the carbon dioxide is cooled for more efficient compression in the compressor.

The simulation models allowed the investigation of advanced coal-fired TPP with different carbon dioxide capture systems.

4 Modeling results

At the first stage, the analysis was carried out for the IGCC without carbon dioxide capture. This power plant’s net efficiency is 53.5% and the carbon dioxide emission is 817 gm/kWh, while combined cycle power plants fired with natural gas without carbon capture have a net efficiency equal to 63.5% according to [39]. The rather low coal-fired power plant efficiency is due to the remarkable losses of 5.8% in the gasification block and the high auxiliary electricity requires for the oxygen production by ASU of 5.9% and its compression upstream the combustion chamber of 1.9%. The high CO\(_2\) emissions for the IGCC are due to the high carbon content. Coal fuel has a 20% higher carbon content compared to natural gas fuel, where emissions are below 450 g/kWh.

At the next stage of the thermodynamic analysis, the effect of carbon capture and storage system on the IGCC energy and environmentally safe performances was estimated. The post-combustion CO\(_2\) capture allows the carbon dioxide emission reduction down to 90 g/kWh. In this version, the TPP net efficiency is as low as 46.6%. The lower efficiency of the coal-fired TPP compared to the IGCC without CO\(_2\) capture is due to the power losses for CO\(_2\) capture. In turn, the natural gas combined cycle power plants with post-combustion CO\(_2\) capture are more efficient compared to the IGCC with post-combustion CO\(_2\) capture due to the lack of gasifica-
According to the research [39], the net efficiency of these plants could achieve 56.9%.

The results of the thermodynamic analysis of the IGCC with pre-combustion carbon capture (case 1) show that the efficiency of this power plant without the utilization of the heat produced in the fuel preparation is 42.3%. Utilization of the water gas shift reactor heat allows higher efficiency of 45.2% (case 2). Utilization of the reactor and the syngas cooler heat (case 3) flows results in the efficiency increase up to 46.5% and utilization of the reactor syngas' and carbon dioxide coolers' thermal power up to 47.2% (case 4). The last case has the carbon dioxide emission of 103 g/kWh, which is 13 g/kWh higher than in the carbon dioxide capture from the exhaust gas. This is due to the large syngas consumption with its binding and the further separation from it of the combustible carbon monoxide.

Part of the heat produced in the fuel preparation in the TPP with pre-combustion capture heat balance is remarkably large so its utilization considerably influences the TPP efficiency improvement. For example, the thermal power emitted in the reactor and coolers is about 23% of the coal combustion heat power. Utilization of this heat in the HRSG reduces the concerned losses to 5% (Fig. 4). The HRSG thermal power increase allows the 5% steam turbine power increase and its condenser heat losses become

![Figure 4: Heat balance of IGCC with pre-combustion carbon capture without heat recovery (case 1) and with heat recovery (case 4).](image-url)
by 13% smaller. Thus, the utilization of the fuel preparation heat increases the cycle net efficiency by almost 5%.

The coal gasification oxy-fuel combustion power cycle has a net efficiency of 36.3% which is the lowest among the low emission power production cycles. This low thermal efficiency is mostly due to the turbine inlet parameters lower than those of the combined cycle. Syngas must be cooled before its compression and supplied to the combustion chamber, but about 10.9% of the cycle consumed power is lost in the syngas cooling after the gasification block. Moreover, the oxygen production by ASU and its compression before combustion consume about 19% of the turbine power production (Fig. 5), which also reduces the net efficiency by the larger TPP auxiliary electricity requirements. The net efficiency of the natural gas-fired oxy-fuel combustion power plants is almost 11% higher according to the modeling results presented in [40] due to zero energy consumption for gasification.

Figure 5: The distribution of power generated by the turbine of the oxy-fuel combustion power plant.

Figures 6, 7, and 8 demonstrate evaluation of thermal efficiency and environmental safety of the reviewed above advanced TPPs with coal gasification and carbon capture and storage systems. The maximal efficiency of 47.2% is reached in a coal-fired combined cycle TPP with the carbon dioxide capture from syngas and utilization of the heat produced in the syngas preparation. In the case of CO₂ capture from exhaust gases, the net efficiency is 46.6%. The coal-fired oxy-fuel combustion power cycle has the
The smallest efficiency of 36.3%. In turn, the oxy-fuel combustion power plant has a minimum carbon dioxide emission of 9 g/kWh.

Table 3 summarizes the thermal efficiency and environmental safety performance in different coal- and natural gas-fired TPPs. The coal- and natural gas-fired combined cycle power plants have the highest thermal efficiency. On the other side, the oxy-fuel combustion power plant has the best environmental safety with carbon dioxide emissions below 10 g/kWh.

Table 3: Energy and environmental indicators of various types TPPs.

<table>
<thead>
<tr>
<th>TPP type</th>
<th>Natural gas</th>
<th>Coal</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td>Initial temperature, ▒C</td>
<td>Net efficiency, %</td>
</tr>
<tr>
<td>Steam turbine plant</td>
<td></td>
<td></td>
</tr>
<tr>
<td>with carbon capture (post-combustion capture)</td>
<td>600–650</td>
<td>47.0</td>
</tr>
<tr>
<td>with carbon capture (post-combustion capture)</td>
<td></td>
<td></td>
</tr>
<tr>
<td>Combined cycle power plant</td>
<td></td>
<td></td>
</tr>
<tr>
<td>without carbon capture</td>
<td></td>
<td>63.5</td>
</tr>
<tr>
<td>with carbon capture</td>
<td></td>
<td></td>
</tr>
<tr>
<td>post-combustion capture</td>
<td>1600–1700</td>
<td>56.9</td>
</tr>
<tr>
<td>pre-combustion capture (case 1)</td>
<td>–</td>
<td>–</td>
</tr>
<tr>
<td>pre-combustion capture (case 4)</td>
<td>–</td>
<td>–</td>
</tr>
<tr>
<td>Oxy-fuel combustion power plant</td>
<td></td>
<td></td>
</tr>
<tr>
<td>with carbon capture (oxy-combustion capture)</td>
<td>1100</td>
<td>47.0</td>
</tr>
</tbody>
</table>

A measure of the energy cost related to CO₂ capture was estimated by the specific primary energy consumption for CO₂ avoided (SPECCA), which is defined as in [41]. According to the research [41], SPECCA for the IGCC with pre-combustion carbon capture is 3.71 MJ/kgCO₂. In this paper,
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Figure 6: The distribution of thermal power supplied to the TPPs with coal gasification and CCS system.

Figure 7: Net efficiency of TPP with coal gasification and low harmful emissions.

Figure 8: Specific CO$_2$ emissions of TPP with coal gasification and low harmful emissions.
SPECCA is 2.52 MJ/kg\textsubscript{CO2} for IGCC with pre-combustion carbon capture (case 1), but for the IGCC with pre-combustion carbon capture (case 4), it is equal to 1.26 MJ/kg\textsubscript{CO2}. Such a low energy consumption for CO\textsubscript{2} removal and high SPECCA value can be explained by the presence of regeneration in case 4 and higher value of initial temperature, which in itself leads to an increase in net efficiency of a power plant.

5 Conclusions

The mathematical simulation models for the TPP with coal gasification and carbon capture and storage systems allowed assessment of the TPP with pre- and post-combustion capture and oxy-fuel combustion power and financial performance, were developed.

The thermal efficiency of the integrated gasification combined cycle with pre-combustion CO\textsubscript{2} capture is 47.2\% which is 0.6\% higher than the post-combustion one and 9.9\% higher than the oxy-fuel combustion. The pre-combustion capture net efficiency is remarkably influenced by the utilization of the heat produced by the water gas shift reactor and the capture system coolers which is about 23\% of the cycle thermal power.

The carbon capture and storage system in a combined cycle TPP with coal gasification may reduce harmful emissions from 817 to below 110 g/kWh and application of the oxy-fuel combustion power cycles to below 10 g/kWh.

The combined cycle TPP with coal gasification and carbon dioxide capture and storage systems have higher power production efficiency of 46.6–47.2\% than the 38.5\% in the coal-fired supercritical steam turbine TPP equipped with CCS which is due to the higher mean-integral heat supply temperature in the combined cycle.

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